only transport, from the higher country to the sea, a mass of solid matter equal to that borne down" by these two rivers. Such an accession of earth would cover annually 1,650 square miles of surface-or, in one year, one third more than the dry land of Rhode Island; in three years, nearly the area of Connecticut; and in twenty-eight years, nearly that of the State of New York, with a layer of soil one foot in thick ness! And this amount is denuded from the water shed of but two rivers! "But," says the unconvinced reader, "how small is the area of New York State when compared with the vast extent of country drained by these mighty streams The foot in New York State must be reduced to a fraction of an inch over the slopes of the Himalayas, and of Northern India." To which we reply, how short a time is twenty eight years compared to the age of these rivers! For on this point other evidence steps in, and we learn that the deposits in their delta, even as far as our limited knowledge of them goes, are sufficient to cover our State with seven hundred feet of earth; or, in other words, that material enough to form a mountain range nine hundrea miles in length twenty-five miles in breadth, and sloping from the plain to a hight of twenty-eight hundred feet, has been in the course of time removed from the basins of the Ganges and the Brah mapootra. Should the reader figure this out he will say 'At this rate you give these rivers an antiquity of twenty housand years." And why not? Or twice as long, if you will? Lyell, with very good grounds for the statement, says of the Mississippi, that it has been transporting its earthy burden to the ocean during a period far exceeding perhaps one hundred thousand years. Perchance, now, you begin to understand why men remained so long in ignorance of the vast operations of Nature? As long as the world was thought to be but six thousand years old, men saw no purpose in her low movements, and the results she had already achieved were but so many incomprehensible puzzles.

## SCIENTIFIO INTELLIGENCE.

## COLORED CEMENTS.

Professor Bottger prepares cement of diverse colors and great hardness by mixing various bases with soluble glass. Soluble soda glass of $33^{\circ}$ B. is to be thoroughly stirred and mixed with fine chalk, and the coloring matter well incor porated. In the course of six or eight hours a hard cement will set, which is capable of a great variety of uses. Bottger recommends the following coloring matters:

1. Wen sifted sulphide of antimony gives a black mass, which, after solidifying, can be polished with agate, and then possesses a fine metallic luster.
2. Fine iron dust, which gives a grey black cement.
3. Zinc dust. This makes a grey mass, exceedingly hard, which, on polishing, exhibits a brilliant metallic luster of zinc, so that broken or defective zinc castings can be mended and restored by a cement that might be called a cold zinc casting. Itadheres firmly to metal, stone, and wood.
4. Carbonate of copper gives a bright green cement
5. Sesquioxide of chromium gives a dark green cement.
6. Thénard's blue, a blue cement.
7. Litharge, a yellow.
8. Cinnabar, a bright red
9. Carmine, a violet-red.
The soluble glass with fine chalk alone gives a white cement of great beauty and hardness.
Sulphide of antimony and iron dust, in equal proportions, stirred in with soluble glass, afford an exceedingly firm black cement; zinc dust and iron in equal proportions yield hard, dark grey cement.
As soluble glass can be kept on hand in liquid form, and the chalk and coloring matters are permanent and cheap, the colored cements can be readily prepared when wanted, and the material can be kept in stock, ready for use, at little expense. Soluble glass is fast becoming one of our most important articles of chemical production.
use of iodine in the mandfacture of chloral.
The enormous consumption of the hydrate of chloral as an anodyne and the expense of its mantafacture, render any modification of the old process of its preparation very acceptable. F. Springmuhl, assistant in the laboratory of Bres lau, proposes the employment of iodine as an improvement To every half pound of alcohol he adds half a grain of
iodine. The alcohol, which is colored brown by the iodine iodine. The alcohol, which is colored brown by the iodine,
soon becomes clear on passing chlorine gas through the mixsoon becomes clear on passing chlorine gas through the mix-
ture, and the hydrochloric acid produced by the decomposition of the alcohol is passed through water for its absorption; while the residue of the vapor is removed by sulphuric
acid and chloride of calcium. The liquid becomes hot at acid and chloride of calcium. The liquid becomes hot at first, and has to be cooled; it is afterwards heated to ebulli-
tion. After passing chlorine gas for twelve hours through the half pound of alcohol contained in a tubulated retort, na more hydrochloric acid is observed, and only pure chlorine gas passes over. The liquid in the retort is neutralized with caustic lime, filtered and distilled. At $161^{\circ}$ Fah., all the iodide of ethyl goes over; and between $230^{\circ}$ and $240^{\circ}$ Fah., the chloral, which is separately condensed, is then mixed with concentrated sulphuric acid, once more distilled, and finally purified by sublimation. The hydrate of chlaral obtained in this way amounted, in two experiments, to ninety and ninety-six per pent of the thearetical quantity, and was af the best quality and free from iadine.
It is said that the purification of the hydrate of chloral can be best accomplished by the use of chloroform, benzale, oil of turpentine, or bisulphide of carbon, as solvents.
turpentine, or bisulphide of carbon, as solvents, 1 part of the hydrate of chloral be dissolved in 5 ar 6 parts of the oil of turpentine at hetween $86^{\circ}$ and $104^{\circ}$ Fah. 6 and the liquid be slowly cooled, beautiful plates and tahles
separate, The best solvent is the bisulphide of carbon; at
$60^{\circ}$ Fah., 1 part of the hydrate of chloral is soluble in 45 parts of the bisulphide; but at temperatures below the boiling point of the solvent, 4 or 5 parts of the bisulphide are sufficient to 1 part of the chloral. By allowing the liquid to cool lowly, large prisms, sometimes an inch long, separate, and in the air rapidly lose all traces of the bisulphide. When prepared in this way, the perfectly pure hydrate of chloral uses between $120^{\circ}$ and $127^{\circ}$ Fah.
For medicinal purposes only the pure, crystalline product ought to be employed.

## artificial alizarine.

One part of anthracen is boiled for a few minutes with 4 o 10 parts of concentrated sulphuric acid diluted with water, and neutralized with carbonate of lime, or with a carbonate of soda or potash; and the sulphates of these bases removed by filtration or crystalization. The resulting liquid is heated to from $356^{\circ}$ to $500^{\circ}$ Fah., with caustic potash, to which chlorate of potash or saltpeterin an amount equal to which chlorate of potash or saltpeter in an amount equal to
the anthracen employed Has been added, so long as a violetthe anthracen employed Ras been added, so long as a violet-
ble color is produced. From this product the alizarin is thrown down by acids.

## are minerals.

Professor Rammelsberg, of Berlin, has recently analyzed two rare minerals, called Fergusonite and Tyrite, the former from Sweden, and the latter from Norway, the composition of which discloses substances so little known that it is diffiof which discloses substances so little known that it is diffi-
cult to see to what uses they could be applied, even if we had them in great abundance. It so often happens, however, that elements of rare occurrence eventually become the very corner stone in some new technical discovery, that it is never
well to pass over any of them as of no value. We give well to pass over any of them as of no value. We give
below the constituents of the minerals, and doubt if many of our readers are familiar with the earths mentioned:

| Tantalic acid | Fergusonite. <br> ... $8 \cdot 73$ | $\begin{aligned} & \text { Tyrite. } \\ & 55.00 \end{aligned}$ |
| :---: | :---: | :---: |
| Columbic acid. | .4016 |  |
| Stannic acid. | 0.91 |  |
| Tungstic acid. | . $30 \cdot 45$ | $30 \cdot 00$ |
| Ceria........ |  | 5.74 |
| Lanthana. | , 7•80 | 3 3.51 |
| Didymia. | ... 4.09 | \} $\begin{aligned} & 3.48\end{aligned}$ |
| Urania. | ... $1 \cdot 98$ | 6.52 |
| Lime... | ., 3•40 | $2 \cdot 36$ |
| Alumina. |  | 1.05 |
| Water. | ... 4477 | $4 \cdot 88$ |
|  | $101 \cdot 99$ | $100 \cdot 54$ |

The Insulation of Telegraph Wires in cities.
Glass, when placed in the shade, becomes completely coated with a thin film of water whenever the moisture contained in the atmosphere amounts to above 40 per cent of saturation. During rain the atmosphere sometimes reaches the point of complete saturation, or 100 per cent. When this is the case, any article of glass, even if exposed to the atmosphere alone, and not to the direct action of the rain, is soon completely covered with moisture, and under these cir cumstances its surface becomes a conductor of electricity.
The atmosphere of all large cities is heavily charged with soot, smoke, and ammoniacal salts, arising from combustion; and these, being taken up by the particles of falling rain and moisture, increase the conducting power of the latter to an enormous extent. Careful experiments made in Manchester, England, where the atmosphere is very impure, showed that the conducting power of the rain water which fell in that city-was more than 300 times that of distilled or absolutely pure water. Speaking of this subject, Latimer Clark says: "Pure water offers a very high resistance, but if it contain any acids or saline matters in solution, the resist ance is mueh smaller; hence it is that clear rain in the coun try does not greatly injure the working of a line, but in towns, where the atmosphere is less pure, the insulation of ten becomes very imperfect in wet weather."
The comparative insulation of wires, in the city and country, under otherwise similar conditions, may be seen by the following actual measurements, taken at the New York office of the Western Union Company: No 1 wire east showed a mileage insulation, between 145 Broadway and Harlem river, of 66,000 ohms, while from Harlem river to New Haven, Conn., the same wire gave 282,000 ohms per mile No. 3 east, to Harlem, gave 53,500 per mile ; Harlem to Hartford, Conn., 218,000 . The insulation in the country exceede that in the city in the proportion of more than 4 to 1.
The European telegraphic engineers have endeavored to surmount this difficulty by changing the insulators at shor intervals, as their surfaces became smoked and dirty. This, however, is but a partial remedy, as the trouble arises as much from the great conductivity of rain water, under the conditions referred to, as it does from dirt upon the surface of the insulators. They have also largely resorted to the expedient of running the wires underground, a method involving great expense, and yet of rather questionable henefit, as far as immunity from interruption is concerned. Considerable embarrassment is also occasioned by inductive ac tion, when underground wires are employed, especially in warking autamatic ar printing instruments.
It is to an American inventor that the credit is due of being the first to discover a practical and effectual means of insulating wires in cities; and equal credit should be accorded to the American telegraphic superintendent wha had the boldness to put the plan into practice on a harge scale, and with the most successful results-we refer to the mag-
nificent lines built by General Anson Stager, of the Western Union Company, in the principal Western cities, which are considered by competent judges to be, perhaps, the finest examples of telegraphic construction in the world.

The hight of the city poles above the ground is sixty-five feet. They carry fifty, No. 9 wires, arranged upon nine cross
arms, and insulated with the Brooks insulator. A test of these lines in red lation, within eight miles from the office, to be so high as to be beyond the range of measurement of either the Siemens universal galvanometer or the Varley differential-the instrument usually employed for these tests. These lines, as specimens of telegraphic engineering, are equally creditable in a mechanical point of view. The massive spars, ranged with mathematical accuracy for miles along the straight and level streets of Chicago, instead of detracting from the ap-
pearance of the thoroughfares, are a positive ornament to pearance of the thoroughfares, are a positive ornament to them. The ordinary sized poles are twenty-one feet in the Central Pacific Railway line, the Michigan Central, and the Philadelphia and Reading Railroad line. The latter, by the way, is a very good specimen of substantial construction, eight wires being carried upon two cross arms, and not high enough from the ground to strain the poles too mich upon the sharp curves which abound upon that road. -The Telegrapher.

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3.

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stopped by the springing of timber, turning of logs, and other causes that stopped by the springing of timber, turning of logs, and other causes tha practical sawyers know. My opinion is that ifE. O. T. try it he will some
day find his mill a wreck. I would also state that I have a day find his mill a wreck. I would also state that I have a gear of his
description 2 -feet diameter, 5 -inch face, run by water power description turns in a minute, used with belt for driving a 48-inch saw. A. O. B., of Vt .

Cement.-F. P. B. can make.a cement for fastening leather to iron or glass, as follows: To 1 quart of glue, after it is dissolved in good when it is fit for use.-O. L. C., of N. H.
Turning Lathe.-If M. C. R. will take a light cut from the bottom of the tail-stock, his lathe will turn true. The tall stock is evi
J. M. D.-The object of our query column, and column of answers to correspondents, is to benefit ourreaders at large, not individual
readers. If you will send the recipes of which you speak we will publish them, but do not intend to make our office a medium of intercommunication on private business matters. The action of a steel magnet or any other magnet, will not render the air magnetic. A machine kept in motion by the attractive force of a permanent magnet would be a per petual motion in the same sense as one kept in constant motion by the action of gravity. A water wheel placed in a never. failiurg stream is a perpetual motion in this sense. What is sought for is, however, a machin
that will move itself independently of static force. Have you got such a machine? If so, we shall be glad to be introduced to it
B. M. \& Co., of Ind.-You are on the right track. By admit ting air behind the bridge wall in the manner proposed, you will probably
consume your smoke. We belleve that heated air, ifforced in under press consume your smoke. We believe that heated air, ifforced in under press.
ure, is better than cold air. If, however, it go in only under ordinary ure, is better than cold air. If, however, it go in only under ordinary
pressure, what you gain by increase of temperature will be, in great measure, lost by expansion, less oxsgen entering in proportion to volume than when it enters cold.
J. A. H., of Ga.--There is no such substance as that you seek. The experiment you propose indicates that you do not understand the frrst principles of electrical science. Better get some good treatise, and inform yourself, than waste time and money in
by any possibility teach you.anything.
M. Y., of Ga.-We shall be glad to hear from you on the subject proposed, but cannot, of course, promise publication till we read your
manuscript. The proportions for Babbitt metal, and method of making the manuscript.; The proportions for Babbitt metal, and method of making the alloy are as follows: Melt 4 parts of copper, and add by degrees 12 parts o best Banca tin, and 8 parts of
melted add 12 parts more of tin.
B. J. of Pa.-Rosner, a Danish Astronomer, first determined the velocity of light in 1675, by observing the eclipses of Jupiter's moons.
It seems to require no time at all to pass over any distance of earth; the It seems to require no time at all
flash seems to be instantaneous.
E. M. F., of N. J.-You may use soda ash in your boiler to remove scale without any danger of hurting the boiler. In some cases it
will loosen the scale, in others it will not. It will do no harm to try it. G. F. C., of -.-Plaster of Paris is prepared for taking casts by simply mixing it with water to the consistence of cream. The mixing must be done rapidly, or it will set before it can be poured into the mold.
0. W. Y. of Conn.-You will find the information you seek in an article on "Artificial Stone," page 263, Vol. Xxiil. of the Scievtific L. R., of N. H.- The motive powers of streams, flowing equalvolumes of water, will be directly as their falls. If a stream through Which a given volume, at a given point, falls ten feet, produce at that point one hundred horse power, the same volume falling at another point
twenty feet would yield two hundred horse power. The horse power of twenty feet would yield two hundred horse power. The horse power of any body of falling water, is the weight in pounds which falls per minute,
multiplied into the distance in feet through which it falls, and the product divided by 33,000 .

## Querits.

[We present herevoth a series of inquiries embracing a variety of topics of Ireater or less general interest. The questions are simple, it is true, but wo
prefer to elicit practical answers from our readers, and hope to be able to make this column qfinquiries and answers a popular and useful feature of the paper.]
1.-Cement for Leather that will Resist Water and Heat. -I wish a cement for leather that will resist the action of water and
2.-Filter for Cisterns.-I see some of your correspon-
nts recommend a wall of soft-burnt bricks for cistern filters. should the all be laid up with mortar or cement, or simply with the bricks alone? A. K.
3.-How can I render scrap lead (such as accumulates in a umber's shop) as soft and tough as pure sheet lead or pig lead? I desire lead, but the scrap is too hard and brittle. Cheapness is of course an ob ead, but the F
eH. J.
4.-Hardening Cast Iron.-How can cast iron be hardened after it is itted and finished, without injury to the finished surface, and
an

